

SYNTHESIS AND TUBERCULOSTATIC ACTIVITY OF NEW BENZIMIDAZOLE DERIVATIVES

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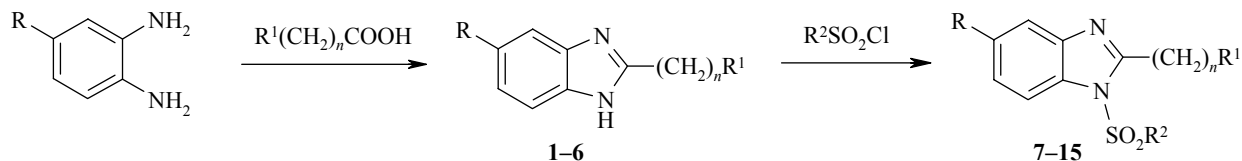
The reactions of *o*-phenylenediamine and 3,4-diaminotoluene with such acids as 3-cyclohexylpropionic, 4-phenylbutyric, 4-cyclohexylbutyric, 3,3-diphenylpropionic, and 3,4-dimethoxyphenylacetic resulted in the formation of the corresponding 2-substituted benzimidazoles. These compounds were transformed into methane- and benzenesulfonamide derivatives. The benzimidazole derivatives obtained were tested *in vitro* for their tuberculostatic activity. Compounds with good activity (MIC 6.2-25 µg/ml) have been found.

Keywords: benzimidazoles, methane- and benzenesulfonamides, tuberculostatics.

Among benzimidazole derivatives, the 2-substituted ones that have varied pharmacological activity (e.g. antitumor, antiverminous, antiviral, hypotensive, spasmolytic, immunosuppressive, neuroleptic, analgesic, antihistaminic) are particularly abundant [1]. Recently the antibacterial (including tuberculostatic) activity of benzimidazole derivatives was reported too [2-6].

In the present study the syntheses of benzimidazole derivatives with cyclohexylalkyl and aralkyl substituents in position 2 were accomplished. *o*-Phenylenediamine and 3,4-diaminotoluene as well as 3-cyclohexylpropionic, 4-phenylbutyric, 4-cyclohexylbutyric, 3,3-diphenylpropionic, and 3,4-dimethoxyphenylacetic acids were the substrates used.

The Phillips method, *i.e.*, the heating of substrates in NH₄Cl, was tested first, but much better yields of benzimidazoles **1-6** were obtained by fusion of the starting materials at 160-180°C, the process of which was elaborated earlier by our team [7, 8]. In the next step compounds **1-6** reacted with methane- and benzenesulfonyl chloride in anhydrous dioxane in the presence of triethylamine, giving the corresponding sulfonamide derivatives **7-15** (Tables 1 and 2).



1-5, 7-13 R = H, **6, 14, 15** R = Me, **1, 3, 6, 7, 10, 11, 14, 15** R¹ = *cyclo*-C₆H₁₁,
2, 8, 9 R¹ = Ph, **4, 12** R¹ = CHPh₂, **5, 13** R¹ = 3,4-(MeO)₂C₆H₃, **7, 8, 10, 14** R² = Me,
9, 11-13, 15 R² = Ph; **1, 7** n = 2, **2, 3, 6, 8-11, 14, 15** n = 3, **4, 5, 12, 13** n = 1

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TABLE 1. Characteristics of the Synthesized Compounds **1-15**

Com- pound	Empirical formula	Found, % Calculated, %			mp, °C, solvent	Yield, %
		C	H	N		
1	C ₁₅ H ₂₀ N ₂	78.81	8.79	12.11	193-194 MeOH	22
		78.90	8.83	12.27		
2	C ₁₆ H ₁₆ N ₂	81.09	6.65	11.97	148-149 Benzene	45
		81.32	6.82	11.85		
3	C ₁₆ H ₂₂ N ₂	79.15	9.32	11.45	151-153 Dioxane	56
		79.29	9.15	11.56		
4	C ₂₁ H ₁₈ N ₂	84.42	6.15	9.15	198-200 Benzene	41
		84.53	6.08	9.39		
5	C ₁₆ H ₁₆ N ₂ O ₂	71.37	5.92	10.27	158-160 Benzene	69
		71.62	6.01	10.44		
6	C ₁₇ H ₂₄ N ₂	79.38	9.28	10.82	143-145 Cyclohexane	47
		79.64	9.44	10.93		
7	C ₁₆ H ₂₂ N ₂ O ₂ S	62.64	7.35	9.02	106-107 MeOH	61
		62.72	7.24	9.14		
8	C ₁₇ H ₁₈ N ₂ O ₂ S	64.75	5.57	8.65	115-118 Cyclohexane	40
		64.94	5.77	8.91		
9	C ₂₂ H ₂₀ N ₂ O ₂ S	70.22	5.17	7.28	120-124 MeOH	54
		70.19	5.35	7.44		
10	C ₁₇ H ₂₄ N ₂ O ₂ S	63.57	7.41	8.48	103-107 Dioxane	73
		63.72	7.55	8.74		
11	C ₂₂ H ₂₆ N ₂ O ₂ S	69.92	6.89	7.11	125-130 MeOH	46
		69.08	6.85	7.32		
12	C ₂₇ H ₂₂ N ₂ O ₂ S	73.81	4.82	6.14	143-147 EtOH	38
		73.95	5.06	6.39		
13	C ₂₂ H ₂₀ N ₂ O ₄ S	64.72	5.28	6.62	115-117 EtOH	50
		64.69	4.94	6.86		
14	C ₁₈ H ₂₆ N ₂ O ₂ S	64.39	7.97	8.17	136-140 MeOH	66
		64.64	7.83	8.38		
15	C ₂₃ H ₂₈ N ₂ O ₂ S	69.48	7.32	6.89	152-153 MeOH	67
		69.66	7.12	7.06		

The compounds obtained were tested against the standard *Mycobacterium tuberculosis* H₃₇Rv strain as well as two strains isolated from tuberculosis patients: one resistant to isonicotinic acid (INH), ethambutol (ETB), and rifampicin (RFP), the other fully susceptible to the tuberculostatics administered. Tuberculostatic activity was determined *in vitro* by the classical test tube method with Youman's liquid medium containing 10% of bovine serum.

Based on the minimum inhibiting concentration (MIC) values obtained, one may conclude that some of the compounds tested exhibited a high tuberculostatic activity. The most active were compounds **1, 2, 4, 6, 7, 10, 14**, as their MIC values were within the limits of 6.2-25 µg/ml (Table 3). The majority of the active compounds possessed cyclohexylethyl or cyclohexylpropyl substituents in position C(2) of the benzimidazole system. The presence of the sulfomethyl group at the nitrogen atom did not change the activity.

EXPERIMENTAL

Melting points were determined with the Reichert apparatus and are uncorrected. The ¹H NMR spectra were recorded with a Varian Gemini 200 (200 MHz) spectrometer in CDCl₃ and given in the δ scale from TMS (recalculated using CHCl₃ signal).

TABLE 2. ¹H NMR Spectra of Compounds **1-15**

Compound	δ , ppm (<i>J</i> , Hz)
1	0.88-1.82 (13H, m, CH ₂ , C ₆ H ₁₁); 2.96 (2H, t, <i>J</i> = 8, CH ₂); 7.22-7.52 (4H, m, C ₆ H ₅)
2	2.10-2.26 (2H, m, CH ₂); 2.70 (2H, t, <i>J</i> = 8, CH ₂); 2.91 (2H, t, <i>J</i> = 8, CH ₂); 7.11-7.57 (4H, m, C ₆ H ₅)
3	0.80-1.93 (15H, m, (CH ₂) ₂ C ₆ H ₁₁); 2.94 (2H, t, <i>J</i> = 8, CH ₂); 7.21-7.57 (4H, m, C ₆ H ₅)
4	3.66 (2H, d, <i>J</i> = 8, CH ₂); 4.63 (1H, t, <i>J</i> = 8, CH); 7.15-7.47 (14H, m, C ₆ H ₅)
5	3.78 (3H, s, CH ₃); 3.85 (3H, s, CH ₃); 4.21 (2H, s, CH ₂); 4.82 (3H, m, C ₆ H ₅); 7.21 (2H, m, C ₆ H ₅); 7.53 (2H, m, C ₆ H ₅)
6	0.75-2.00 (15H, m, (CH ₂) ₂ C ₆ H ₁₁); 2.43 (3H, s, CH ₃); 2.85 (2H, t, <i>J</i> = 7, CH ₂); 6.82-7.56 (3H, m, C ₆ H ₅)
7	0.96-1.87 (13H, m, CH ₂ C ₆ H ₁₁); 3.11-3.22 (5H, m, CH ₂ and CH ₃); 7.26-7.40 (2H, m, C ₆ H ₅); 7.70-7.87 (2H, m, C ₆ H ₅)
8	2.28 (2H, m, CH ₂); 2.83 (2H, t, <i>J</i> = 7, CH ₂); 3.18 (5H, m, CH ₂ and CH ₃); 7.19-7.88 (9H, m, C ₆ H ₅)
9	2.24 (2H, m, CH ₂); 2.80 (2H, t, <i>J</i> = 7, CH ₂); 3.15 (2H, t, <i>J</i> = 7, CH ₂); 7.11-8.08 (14H, m, C ₆ H ₅)
10	2.86-2.01 (15H, m, (CH ₂) ₂ C ₆ H ₁₁); 3.11 (2H, t, <i>J</i> = 8, CH ₂); 3.22 (3H, s, CH ₃); 7.26-7.88 (4H, m, C ₆ H ₅)
11	0.81-1.98 (11H, m, C ₆ H ₁₁); 2.25 (2H, m, CH ₂); 2.80 (2H, t, <i>J</i> = 7, CH ₂); 3.12 (2H, t, <i>J</i> = 7, CH ₂); 7.20-8.20 (9H, m, C ₆ H ₅)
12	3.94 (2H, d, <i>J</i> = 7, CH ₂); 5.08 (1H, t, <i>J</i> = 7, CH); 7.12-7.99 (19H, m, C ₆ H ₅)
13	3.73 (6H, s, CH ₃); 3.81 (2H, s, CH ₂); 6.51 (3H, m, C ₆ H ₅); 7.26-7.93 (9H, m, C ₆ H ₅)
14	0.85-2.00 (15H, m, (CH ₂) ₂ C ₆ H ₁₁); 2.46 (3H, s, CH ₃); 3.02-3.25 (5H, m, CH ₂ and CH ₃); 7.06-7.75 (3H, m, C ₆ H ₅)
15	0.72-2.11 (15H, m, (CH ₂) ₂ C ₆ H ₁₁); 2.50 (3H, s, CH ₃); 3.15 (2H, t, <i>J</i> = 7, CH ₂); 7.06-7.95 (8H, m, C ₆ H ₅)

TABLE 3. Tuberculostatic Activity of Compounds **1-15**

Compound	MIC $\mu\text{g/ml}$		
	H ₃₇ Rv	Drug-resistant strain	Drug-susceptible strain
1	3.1	6.2	6.2
2	6.2	25	12.5
3	12.5	6.2	25
4	6.2	25	12.5
5	25	50	25
6	6.2	6.2	6.2
7	25	6.2	25
8	25	50	25
9	12.5	50	12.5
10	6.2	25	12.5
11	12.5	25	12.5
12	100	100	100
13	25	50	25
14	12.5	6.2	6.2
15	25	25	25

Cyclohexylalkyl- and Arylalkylbenzimidazoles 1-6. *o*-Phenylenediamine (10 mmol) or 3,4-diaminotoluene (10 mmol) and the corresponding acid – 3-cyclohexylpropionic, 4-phenylbutyric, 4-cyclohexylbutyric, 3,3-diphenylpropionic, or 3,4-dimethoxyphenylacetic (10 mmol) were heated on Wood's metal bath at 160-180°C for 1 h. On cooling down, 10% NaOH solution (10 ml) was added, and the whole

stirred and allowed to stand for 24 h. Then the precipitated compounds **1-6** were filtered off, washed with water to the neutral reaction, dried, and purified by crystallization.

1-Methane- and 1-Benzenesulfonyl-2-substituted Benzimidazoles 7-15. Compounds **1-6** (5 mmol) were dissolved in anhydrous dioxane (10 ml), treated with triethylamine (2 ml) and then dropwise with methane- and benzenesulfonyl chloride (10 mmol), and the whole stirred while heating at 50°C for 5 h and then at ambient temperature for another 12 h. The solvent was evaporated and some water with ice added. After a few hours the precipitated compounds **7-15** were filtered off, washed with water, and crystallized.

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